

INFLUENCE OF REACTION CONDITIONS OF LIQUEFACTION IN THE VISCOSITY OF KRAFT LIGNIN-BASED POLYOLS

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INTRODUCTION

- Polyols are an important kind of compound due to the presence of hydroxyl groups which can react with other compounds to form a great variety of products.
- However, most of the commercial polyols are based in fossil source.
- In the other side, Kraft lignin has in its structure hydroxyl groups, it is renewable and is an abundant industrial waste.

- The Kraft lignin-based polyol is a potential candidate to replace those of petrochemical source in formulations such as phenolic resin and polyurethane.
- Properties → has a strong dependence of raw material source and of the reaction conditions.
- Viscosity → is one of the properties which is often used as a parameter for the use of a commercial polyol.

MATERIALS AND METHODS

Polyethylene glycol#400 (PEG)

Glycerol (G)

PEG: G
80:20



Catalyst:solvents ratio (0, 3, and 6% H₂SO₄)

Mass:solvents ratios (15, 20, and 25%)

Reaction time (60, 80, and 100 min)

Figure 1 – Reaction conditions of liquefaction under reflux

- The experiments followed an experimental design that totaled 27 runs
- Response Surface Methodology was used to show the influence of independent variables in the behavior of viscosity

Table 1: Experimental design

| Experiment | Time | Catalyst | Mass |
|------------|------|----------|------|
| 01 | 80 | 3 | 25 |
| 02 | 60 | 0 | 20 |
| 03 | 60 | 6 | 25 |
| 04 | 60 | 3 | 20 |
| 05 | 80 | 0 | 20 |
| 06 | 80 | 0 | 25 |
| 07 | 60 | 6 | 15 |
| 08 | 60 | 0 | 15 |
| 09 | 80 | 0 | 15 |
| 10 | 100 | 0 | 15 |
| 11 | 100 | 6 | 20 |
| 12 | 60 | 3 | 25 |
| 13 | 100 | 0 | 20 |
| 14 | 80 | 6 | 20 |
| 15 | 100 | 3 | 25 |
| 16 | 100 | 3 | 15 |
| 17 | 80 | 6 | 15 |
| 18 | 80 | 3 | 15 |
| 19 | 100 | 6 | 15 |
| 20 | 60 | 0 | 25 |
| 21 | 80 | 3 | 20 |
| 22 | 100 | 0 | 25 |
| 23 | 60 | 3 | 15 |
| 24 | 60 | 6 | 20 |
| 25 | 100 | 6 | 25 |
| 26 | 80 | 6 | 25 |
| 27 | 100 | 3 | 20 |

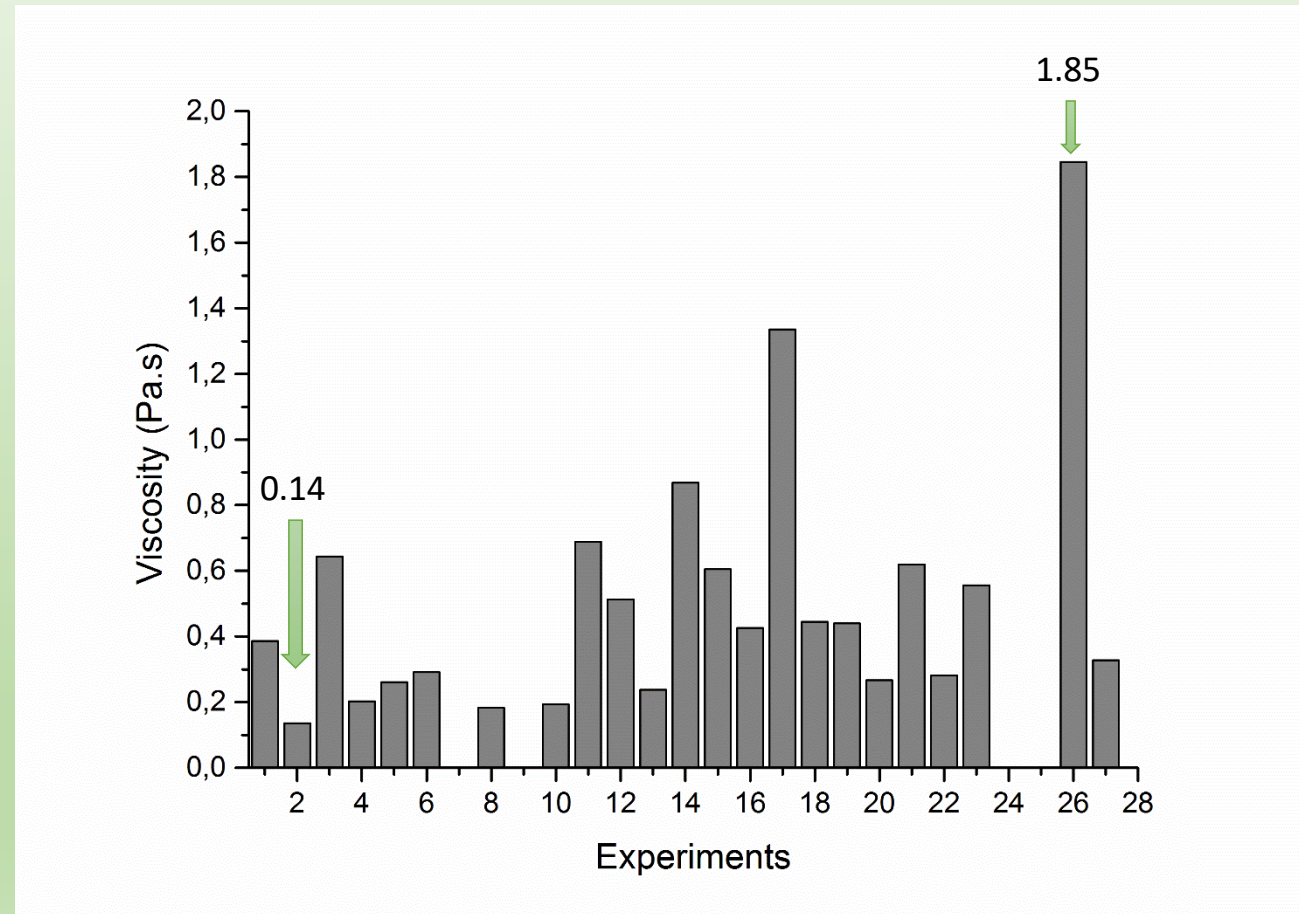


Figure 2 – Viscosity values of liquefied Kraft lignin-based polyol.

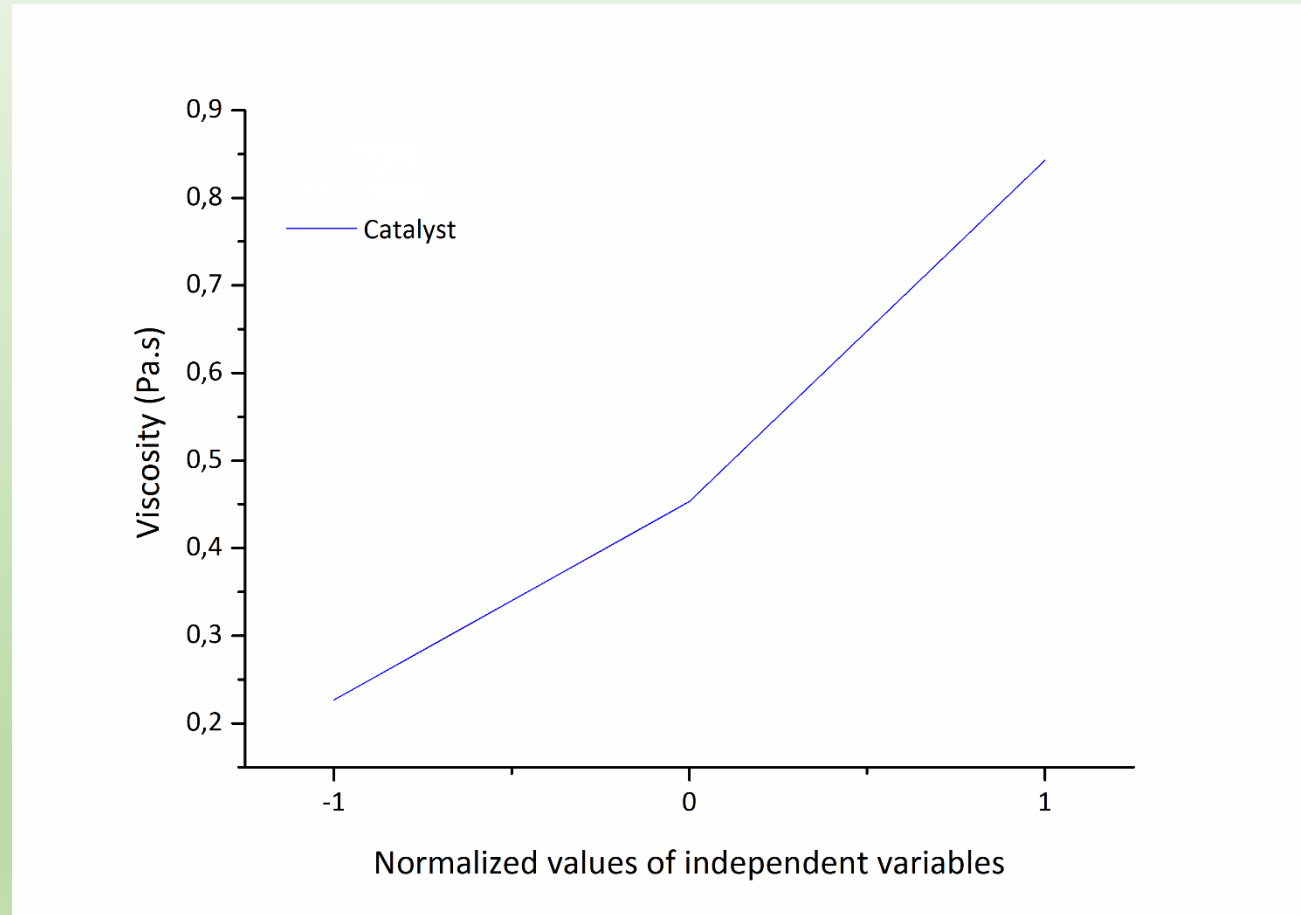


Figure 3 – Average values of viscosity as a function of independent variables

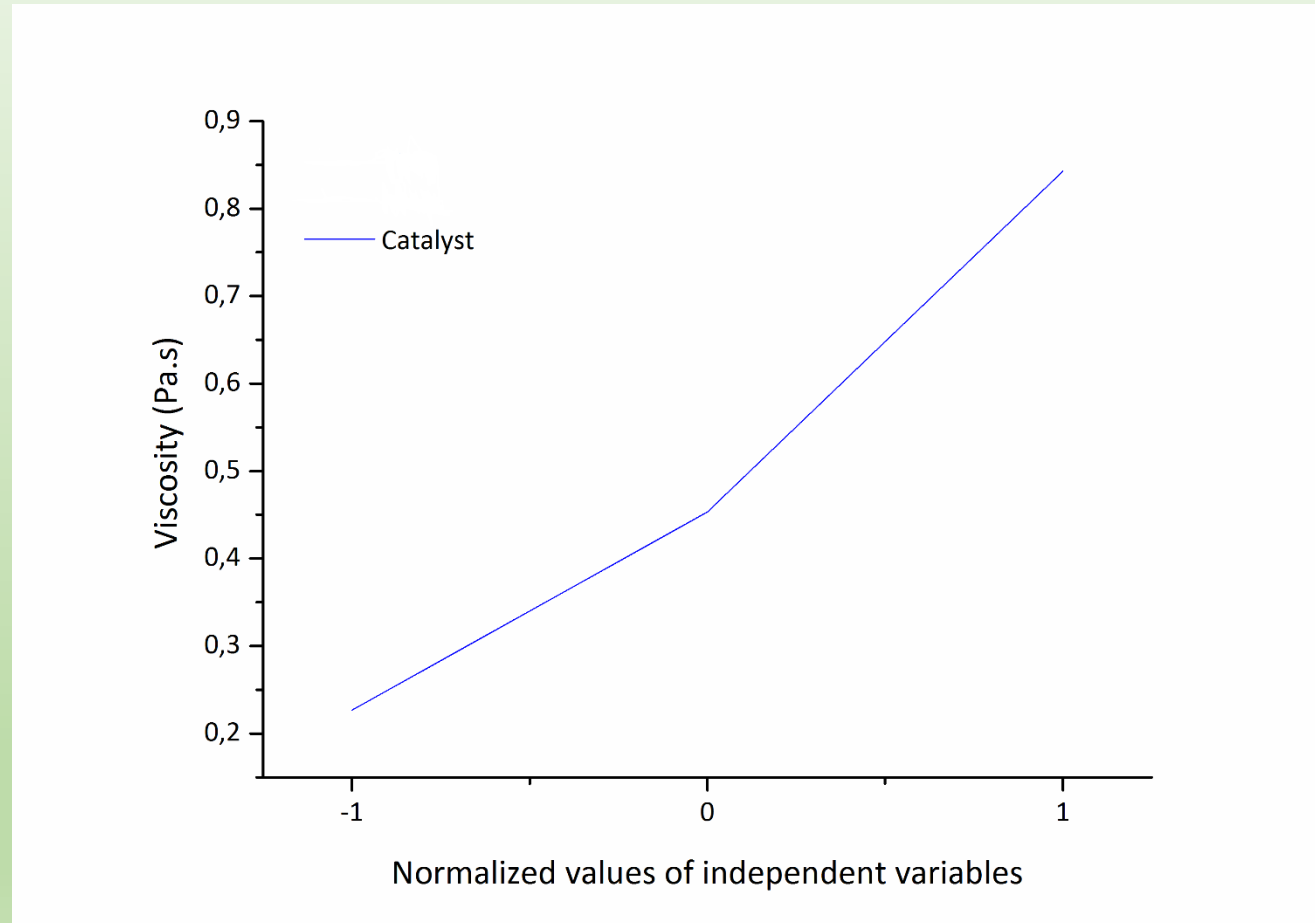


Figure 3 – Average values of viscosity as a function of independent variables

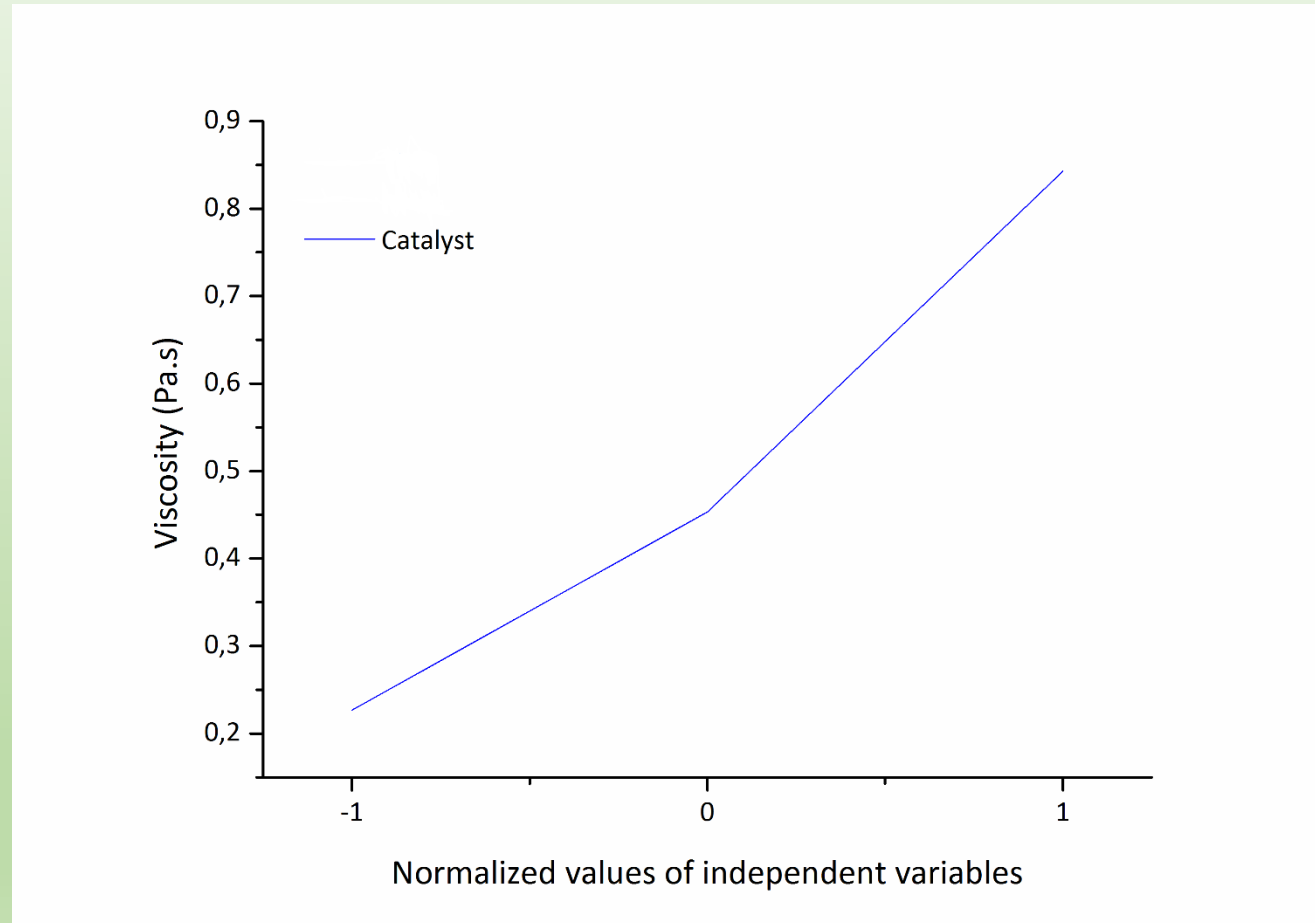


Figure 3 – Average values of viscosity as a function of independent variables

Table 2: Viscosity variance analysis (ANOVA) of liquefied Kraft lignins.

| Source | Sum of squares | Df | Average square | F Rate ^(a) | P value |
|-----------------|----------------|----|----------------|-----------------------|-----------------------|
| Main effect | | | | | |
| A: Catalyst | 1.74765 | 2 | 0.873825 | 13.27 | 0.0029 ^(b) |
| B: Mass | 0.155493 | 2 | 0.0777467 | 1.18 | 0.3553 |
| C: Time | 0.473254 | 2 | 0.236627 | 3.59 | 0.0770 |
| Interactions | | | | | |
| AB | 0.114637 | 4 | 0.0286594 | 0.44 | 0.7801 |
| AC | 0.697839 | 4 | 0.17446 | 2.65 | 0.1122 |
| BC | 0.0280752 | 4 | 0.0070188 | 0.11 | 0.9769 |
| Residue | 0.526674 | 8 | 0.0658343 | | |
| Corrected total | 3.74362 | 26 | | | |

^a Are based on average square residual error, ^b Significant at 95% confidence level

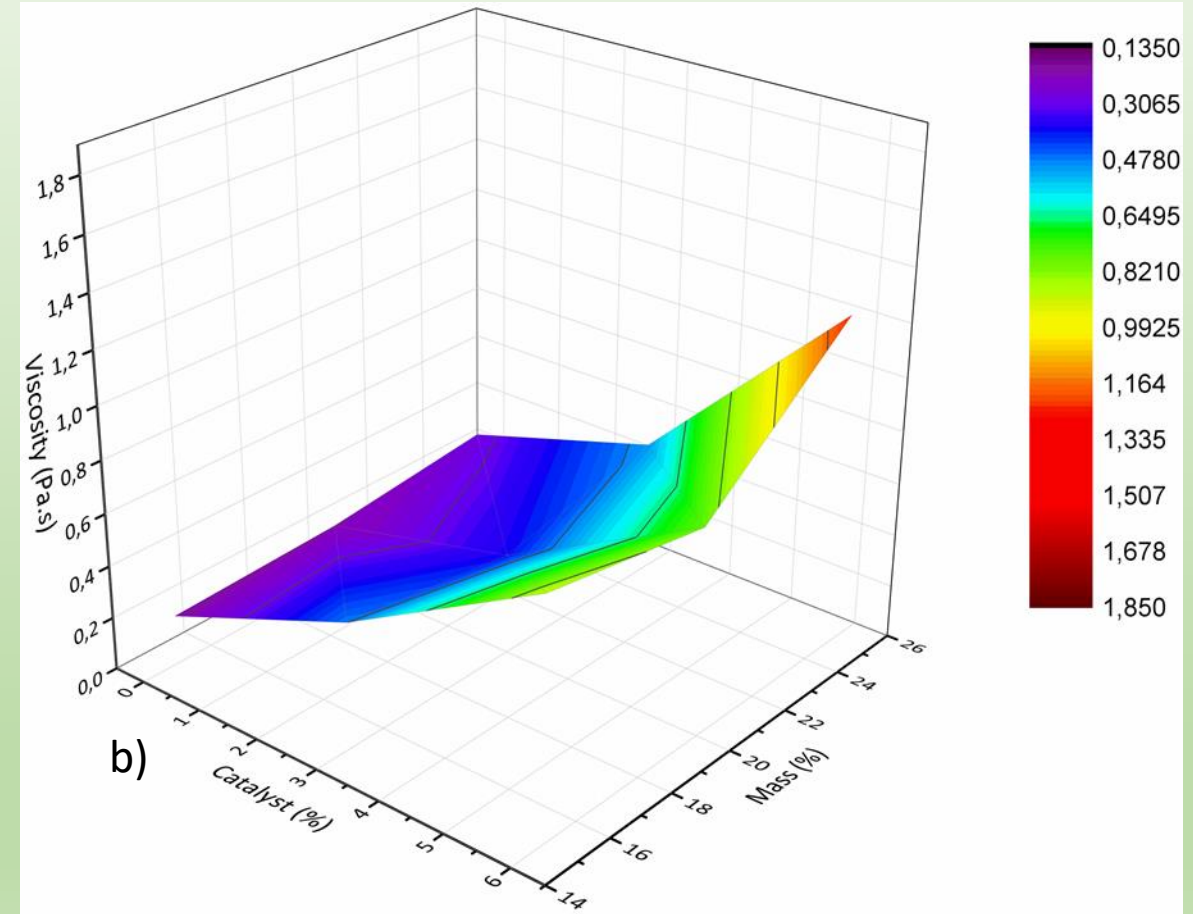
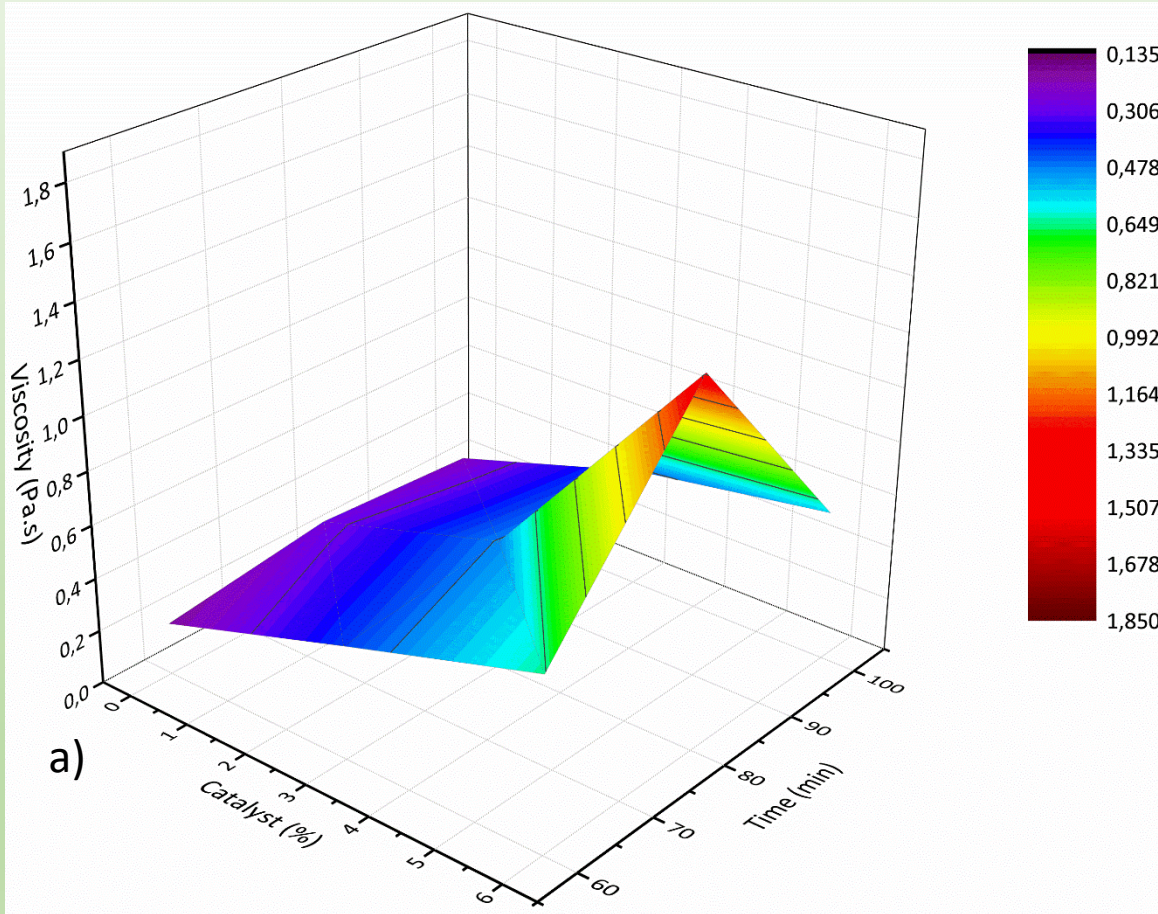


Figure 4: Interaction between variables a) catalyst/time and b) catalyst/mass

CONCLUSIONS

- All Kraft lignin-based polyols have suitable viscosity to replace fossil-based polyols in formulations such as rigid foams
- The lower viscosity values were observed in the absence of sulfuric acid
- The highest viscosities were verified with 6% of sulfuric acid



Thanks for your attention!

